

Supporting Information

**A Spectroscopic Investigation of the Primary Reaction Intermediates in the
Oxidation of Levitated Droplets of Energetic Ionic Liquids**

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Experimental Methods

A single droplet of the ionic liquid was levitated in an ultrasonic levitator enclosed within a pressure-compatible process chamber. The apparatus has been discussed in detail elsewhere.¹⁻² The chamber was evacuated and then filled to 880 Torr with argon (Airgas, 99.9999%) or a gas mixture of 1.3% nitrogen dioxide (Aldrich, $\geq 99.5\%$) and 98.7% argon. The 1-methyl-4-amino-1,2,4-triazolium dicyanamide ([MAT][DCA]) was synthesized as discussed in reference 3. The [MAT][DCA]-capped, hydrogen-terminated boron nanoparticles were produced by high-energy ball-milling boron powder in a hydrogen atmosphere, resulting in hydrogen-terminated nanoparticles, and then capping with [MAT][DCA] to generate air-stable, unoxidized particles, as described in reference 4. A piezoelectric transducer oscillating at 58 kHz generates ultrasonic sound waves that reflect multiple times between the transducer and a concave-shaped reflector to create the standing wave that levitates the droplet. The [MAT][DCA] or [MAT][DCA] with boron nanoparticles was injected into the center of the levitator using a syringe and needle to form a droplet with a diameter of typically 1 mm. The chemical modifications of the levitated droplets were monitored *in situ* by Raman^{1-2, 5} and UV-visible (StellarNet Inc., Silver-Nova) spectrometers. The Raman transitions are excited by the 532 nm line of a diode-pumped, Q-switched Nd:YAG laser. The Raman-shifted photons, which were backscattered from the droplet, are focused by a lens into a HoloSpec $f/1.8$ holographic imaging spectrograph equipped with a PI-Max 2 ICCD camera from Princeton Instruments. In the UV-visible spectrometer, the output from a high-brightness UV-visible source is transmitted via a fiber-optic feedthrough on a 35 CF flange to a reflectance probe located inside of the vacuum chamber. The backscattered light is collected by a read fiber, exits the chamber via a different connector on the same 35 CF fiber-optic feedthrough, and finally enters a UV-visible spectrometer that covers the 190 nm to 1100 nm spectral range. The Fourier-transform infrared (FTIR) spectra were taken before and after the reaction using an attenuated total reflection (ATR) accessory with a Thermo Scientific Nicolet 6700 FTIR spectrometer.¹

Computational Methods

A variant of the quantum-mechanical continuum solvation model SMD,⁶ known as the generic ionic liquid (SMD-GIL) model,⁷ was employed to calculate the vibrational frequencies of the reactants and products in the condensed phase of the ionic liquid. The SMD-GIL calculations were performed using Gaussian 09⁸ with an M06/6-31+G(d,p) basis set. In detail, the SMD-GIL solvent descriptor input parameters used to simulate [MAT][DCA] as the solvent are: $\epsilon = 11.50$, $n = 1.4300$, $\gamma = 61.24$, $\Sigma\alpha_2^H = 0.229$, $\Sigma\beta_2^H = 0.265$, $\phi = 0.2308$, and $\psi = 0.0000$.⁷ After calculating the frequencies, a scaling factor is needed for agreement between experiment and theory. For the M06 level of theory used here, all the vibrational frequencies presented have therefore been scaled by the factor of 0.989.

To interpret the UV-visible spectra in the 200 - 1100 nm wavelength range, we investigated the electronic transitions of each relevant species. Time-dependent density functional theory (TD-DFT) at the M06/6-31 + G(d,p) level was therefore applied to [MAT][DCA], NO₂, and [MAT][DCA] bonded with NO₂ as shown in Figure 1. The calculations were performed using Gaussian09 for the first 50 electronic transitions of each species including singlets and triplets states. The electronic transitions were assigned, for example to $\pi \rightarrow \pi^*$ or $\pi^* \rightarrow \pi^*$, by visual inspection of the molecular orbitals involved in each transition.

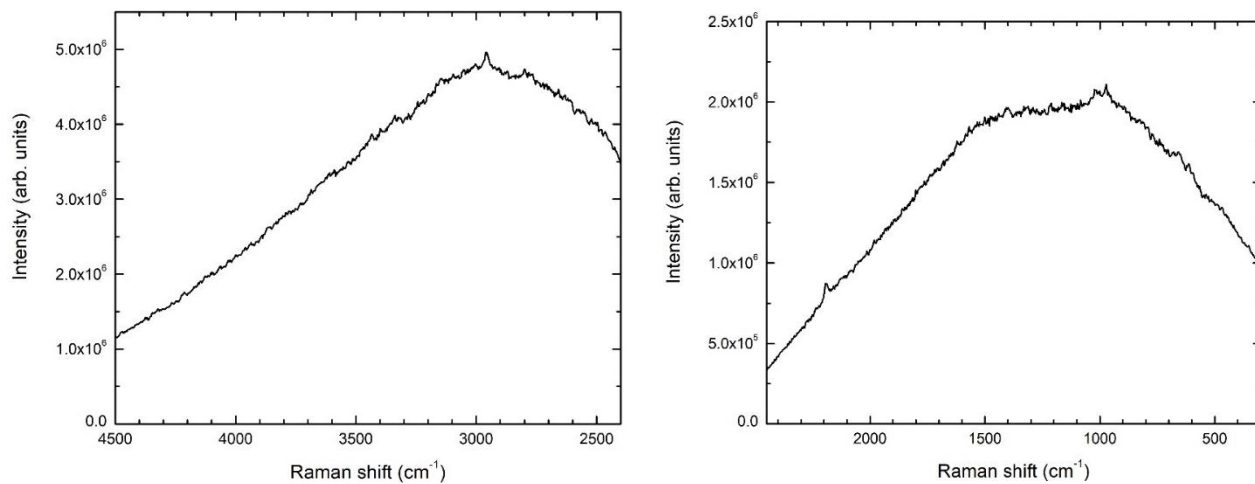


Figure S1. Raman spectra of [MAT][DCA] with hydrogen-capped, boron nanoparticles levitated in argon.

Table S1. Assignments for the observed peaks in the infrared spectra of [MAT][DCA] and [MAT][DCA] doped with hydrogen-capped boron nanoparticles by comparison with GIL theoretical values for [MAT][DCA], the measured values in ^a Ref. [9], ^b Ref. [10], ^c Ref. [11] for the [DCA]⁻ anion, and ^d Ref. [12] for the B-H bond

number	vibrational mode	theoretical and measured reference wavenumbers (cm ⁻¹)	measured wavenumbers [MAT][DCA] (cm ⁻¹)	measured wavenumbers boron-doped [MAT][DCA] (cm ⁻¹)
v ₁	[DCA] ⁻ wag	31		
v ₂	[DCA] ⁻ rock	45		
v ₃	[DCA] ⁻ rock	54		
v ₄	[DCA] ⁻ rock	75		
v ₅	[DCA] ⁻ rock	94		
v ₆	N(CN) ₂ bend [DCA] ⁻	140		
v ₇	CH ₃ rock	173		
v ₈	CH ₃ rotation	196		
v ₉	out-of-plane symmetric NH ₂ , CH ₃ wag	231		
v ₁₀	out-of-plane symmetric NH ₂ , CH ₃ wag 2	290		
v ₁₁	NH ₂ rock	302		
v ₁₂	NH ₂ rotation	335		
v ₁₃	antisymmetric NH ₂ , CH ₃ in-plane wag	450		
v ₁₄	in-plane antisymmetric N-C≡N bend of [DCA] ⁻	514		
v ₂ [*]	in-plane antisymmetric N-C≡N bend of [DCA] ⁻	517 ^{a)}	511	512
v ₃ [*]	out-of-plane antisymmetric N-C≡N bend of [DCA] ⁻	530 ^{a)}	524	524
v ₄ [*]	out-of-plane symmetric N-C≡N bend of [DCA] ⁻	543 ^{a)}	540	540
v ₁₅	out-of-plane symmetric N-C≡N bend of [DCA] ⁻	541		
v ₁₆	out-of-plane antisymmetric N-C≡N bend of [DCA] ⁻	554		
v ₁₇	symmetric N-NH ₂ and N-CH ₃ stretch	613	601	601
v ₁₈	N(4) umbrella	620	613	613
v ₁₉	in-plane symmetric N-C-N bend in [DCA] ⁻	651		
v ₂₀ v ₅ [*]	H ₃ CN(1)-N(2)CH torsion in-plane C-N-C bend of [DCA] ⁻	659 666 ^{b)}	657	657
v ₂₁	antisymmetric ring mode with N-CH ₃ and N-NH ₂ stretch	747	737	737
v ₂₂ v ₂₃	C(3)H, C(5)H antisymmetric out-of-plane wag C(3)H, C(5)H symmetric out-of-plane wag	871 892	880	878
v ₆ [*]	C-N-C symmetric stretch of [DCA] ⁻	904 ^{b)}	905	905
v ₂₄	C-N-C symmetric stretch in [DCA] ⁻	936		
v ₂₅	NH ₂ rock	958		
v ₂₆	N(1)-N(2) stretch, C(3)H in-plane wag	993	978	978

ν_{27}	ring C-N-C symmetric stretch, C(5) wag and CH ₃ rock	1059	1071	1071
ν_{28}	N-C-N ring bend, CH ₃ rock	1085	1091	1091
ν_{29}	CH ₃ rock	1120		
ν_{30}	C(5)H wag, N(1)-CH ₃ stretch	1168	1171	1170
ν_{31}	C(3)H, C(5)H antisymmetric in-plane wag	1197	1212	1213
ν_{32}	ring NN stretch	1287		
ν_7^*	C-N-C antisymmetric stretch of [DCA] ⁻	1309 ^{b)}	1310	1310
ν_{33}	ring C-N-C antisymmetric stretch, NH ₂ rock	1354		
ν_{34}	ring C-N-C antisymmetric stretch, NH ₂ rock 2	1382		
ν_{35}	CH ₃ umbrella	1396		
ν_{36}	C-N-C antisymmetric stretch in [DCA] ⁻	1403		
ν_{37}	CH ₂ wag	1405	1407	1407
ν_{38}	CH ₂ wag 2	1418	1439	1439
ν_{39}	N(2)-C(3) and N(4)-C(5) symmetric stretch	1455	1455	1456
ν_{40}	N(2)-C(3) and N(4)-C(5) antisymmetric stretch	1566	1535	1535
ν_{41}	N(1)-C(5) stretch, N-NH ₂ and N-CH ₃ antisymmetric stretch	1588	1571	1570
ν_{42}	NH ₂ scissor	1638	1630	1630
ν_8^*	C≡N antisymmetric stretch of [DCA] ⁻	2133 ^{b)}	2126	2126
ν_9^*	C≡N symmetric stretch of [DCA] ⁻	2192 ^{b)}	2197	2197
$\nu_6^* + \nu_7^*$	[DCA] ⁻ combination mode	2227 ^{c)}	2237	2237
ν_{43}	[DCA] ⁻ antisymmetric stretch	2232		
ν_{44}	[DCA] ⁻ symmetric stretch	2275		
$\nu(\text{B-H})$	B-H stretch	2565-2480 ^{d)}	-	2494
ν_{45}	CH ₃ symmetric stretch	3038	2971	2971
ν_{46}	CH ₃ antisymmetric stretch	3148	3022	3021
ν_{47}	CH ₃ antisymmetric stretch 2	3162	3080	3079
ν_{48}	C(5)H stretch	3234	3142	3141
ν_{49}	C(3)H stretch	3242	3227	3218
ν_{50}	symmetric NH stretch	3317	3294	3291
ν_{51}	antisymmetric NH stretch	3527	3507	3505

Table S2. Assignments for the observed peaks in the Raman spectrum of [MAT][DCA] by comparison with GIL theoretical values for [MAT][DCA], and the measured values in ^a Ref. [9], ^b Ref. [10], ^c Ref. [11] for the [DCA]⁻ anion.

number	vibrational mode	theoretical and measured reference wavenumbers (cm ⁻¹)	measured wavenumbers [MAT][DCA] (cm ⁻¹)
v ₁	[DCA] ⁻ wag	31	
v ₂	[DCA] ⁻ rock	45	
v ₃	[DCA] ⁻ rock	54	
v ₄	[DCA] ⁻ rock	75	
v ₅	[DCA] ⁻ rock	94	
v ₆	N(CN) ₂ bend [DCA] ⁻	140	
v ₇	CH ₃ rock	173	
v ₈	CH ₃ rotation	196	
v ₉	out-of-plane symmetric NH ₂ , CH ₃ wag	231	
v ₁₀	out-of-plane symmetric NH ₂ , CH ₃ wag 2	290	
v ₁₁	NH ₂ rock	302	
v ₁₂	NH ₂ rotation	335	
v ₁₃	antisymmetric NH ₂ , CH ₃ in-plane wag	450	446
v ₁₄	in-plane antisymmetric N-C≡N bend of [DCA] ⁻	514	
v ₃ [*]	out-of-plane symmetric N-C≡N bend of [DCA] ⁻	530 ^{a)}	532
v ₁₅	out-of-plane symmetric N-C≡N bend of [DCA] ⁻	541	
v ₁₆	out-of-plane antisymmetric N-C≡N bend of [DCA] ⁻	554	
v ₁₇	symmetric N-NH ₂ and N-CH ₃ stretch	613	606
v ₁₈	N(4) umbrella	620	
v ₁₉	in-plane symmetric N-C-N bend in [DCA] ⁻	651	
v ₂₀ v ₅ [*]	H ₃ CN(1)-N(2)CH torsion in-plane C-N-C bend of [DCA] ⁻	659 666 ^{b)}	655
v ₂₁	antisymmetric ring mode with N-CH ₃ and N-NH ₂ stretch	747	729
v ₂₂	C(3)H, C(5)H antisymmetric out-of-plane wag	871	863
v ₂₃ v ₆ [*]	C(3)H, C(5)H symmetric out-of-plane wag C-N-C symmetric stretch of [DCA] ⁻	892 904 ^{b)}	897
v ₂₄	C-N-C symmetric stretch in [DCA] ⁻	936	
v ₂₅	NH ₂ rock	958	971
v ₂₆	N(1)-N(2) stretch, C(3)H in-plane wag	993	1038
v ₂₇	ring C-N-C symmetric stretch, C(5) wag and CH ₃	1059	1059

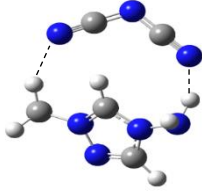
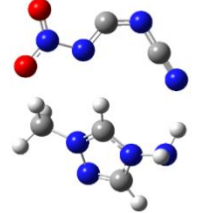
	rock		
ν_{28}	N-C-N ring bend, CH ₃ rock	1085	1077
ν_{29}	CH ₃ rock	1120	
ν_{30}	C(5)H wag, N(1)-CH ₃ stretch	1168	1163
ν_{31}	C(3)H, C(5)H antisymmetric in-plane wag	1197	1206
ν_{32}	ring NN stretch	1287	
ν_7^*	C-N-C antisymmetric stretch of [DCA] ⁻	1309 ^{b)}	1315
ν_{33}	ring C-N-C antisymmetric stretch, NH ₂ rock	1354	
ν_{34}	ring C-N-C antisymmetric stretch, NH ₂ rock 2	1382	
ν_{35}	CH ₃ umbrella	1396	
ν_{36}	C-N-C antisymmetric stretch in [DCA] ⁻	1403	
ν_{37}	CH ₃ umbrella	1405	1402
ν_{38}	CH ₂ wag 2	1418	
ν_{39}	N(2)-C(3) and N(4)-C(5) symmetric stretch	1455	1448
ν_{40}	N(2)-C(3) and N(4)-C(5) antisymmetric stretch	1566	1530
ν_{41}	N(1)-C(5) stretch, N-NH ₂ and N-CH ₃ antisymmetric stretch	1588	1566
ν_{42}	NH ₂ scissor	1638	1624
ν_8^*	C≡N antisymmetric stretch of [DCA] ⁻	2133 ^{b)}	2132
ν_9^*	C≡N symmetric stretch of [DCA] ⁻	2192 ^{b)}	2191
$\nu_6^* + \nu_7^*$	[DCA] ⁻ combination mode	2227 ^{c)}	2236
ν_{43}	[DCA] ⁻ antisymmetric stretch	2232	
ν_{44}	[DCA] ⁻ symmetric stretch	2275	
$2\nu_{37}$	antisymmetric mixed mode	2804	2803
ν_{45}	CH ₃ symmetric stretch	3038	2954
ν_{46}	CH ₃ antisymmetric stretch	3148	3029
ν_{47}	CH ₃ antisymmetric stretch 2	3162	3109
ν_{48}	C(5)H stretch	3234	3160
ν_{49}	C(3)H stretch	3242	
ν_{50}	symmetric NH stretch	3317	3319

Table S3. Theoretical wavenumbers and vibrational mode assignments for [MAT][DCA] bonded with nitrogen dioxide (Figure 1) calculated using the SMD-GIL model at the M06/6-31+G(d,p) level of theory.

number	vibrational mode	theoretical wavenumbers (cm ⁻¹)
v ₁	NO ₂ rock	11
v ₂	anion rock 1	32
v ₃	anion rock 2	44
v ₄	N=C=N wag	65
v ₅	Cation rock	69
v ₆	CH ₃ to NO ₂ stretch	94
v ₇	C-N-CN bend of anion	98
v ₈	O ₂ NN-CNCN torsion	138
v ₉	CH ₃ rotation	159
v ₁₀	CH ₃ rotation and C-N-CN bend	177
v ₁₁	CH ₃ rotation and N(4)C(3)-N(2)CH ₃ torsion	215
v ₁₂	C=N-NO ₂ in-plane bend	224
v ₁₃	NH ₂ wag	291
v ₁₄	NH ₂ wag and N-N-CH ₃ in plane bend	307
v ₁₅	C=N=C=N out-of-plane torsion + NH ₂ rotation	332
v ₁₆	O ₂ N-NC torsion in anion	354
v ₁₇	O ₂ NN=C=N in-plane bend + N-C≡N bend	432
v ₁₈	antisymmetric NH ₂ and CH ₃ in plane wag	456
v ₁₉	N-C≡N out-of-plane bend	570
v ₂₀	NO ₂ rock + N=C=N in-plane bend + C(3)H bend	608
v ₂₁	N-NH ₂ + N-CH ₃ symmetric in-plane bend	611
v ₂₂	C(3) + C(5) out-of-plane symmetric bend	618
v ₂₃	N-C≡N in-plane bend	653
v ₂₄	N(4) + C(5) out-of-plane bend	661
v ₂₅	N-N=C bend + C(5)H in-plane wag	745
v ₂₆	N-NO ₂ umbrella	779
v ₂₇	N=C=N-NO ₂ bend + N-NO ₂ umbrella	784
v ₂₈	C(3)H + C(5)H out-of-plane antisymmetric wag	876
v ₂₉	C(3)H + C(5)H out-of-plane symmetric wag	896
v ₃₀	N-NO ₂ symmetric stretch	946
v ₃₁	N-NO ₂ symmetric stretch + NH ₂ wag	982
v ₃₂	ring NN stretch + NH ₂ wag	996
v ₃₃	N-CN stretch	1029
v ₃₄	C(3)H + C(5)H in-plane wag	1054
v ₃₅	CH ₃ in-plane rocking	1085

V ₃₆	CH ₃ out-of-plane rocking	1125
V ₃₇	C(3)H in-plane wag	1161
V ₃₈	C(5)H in-plane wag	1201
V ₃₉	C=N-NO ₂ stretch	1271
V ₄₀	ring NN stretch + CH ₂ wag + C(3)H + C(5)H in-plane symmetric wag	1288
V ₄₁	N=C=N-NO ₂ symmetric stretch + NH ₂ rock	1351
V ₄₂	ring CNC antisymmetric stretch	1355
V ₄₃	NH ₂ rock	1370
V ₄₄	CH ₃ umbrella	1398
V ₄₅	ring breathing + CH ₃ antisymmetric wag 1	1429
V ₄₆	ring breathing + CH ₃ antisymmetric wag 2	1405
V ₄₇	ring breathing + CH ₃ antisymmetric wag 3	1420
V ₄₈	ring CNNC antisymmetric stretch + C(3)H in-plane wag	1456
V ₄₉	O=N=O antisymmetric stretch	1568
V ₅₀	ring CNNC symmetric stretch + C(3)H in-plane wag	1576
V ₅₁	NH ₂ bend	1585
V ₅₂	O ₂ N-NC=NCN stretch	1632
V ₅₃	N-C≡N antisymmetric stretch	2256
V ₅₄	CH ₃ symmetric stretch	3032
V ₅₅	C(5)H stretch	3145
V ₅₆	CH ₂ antisymmetric stretch	3147
V ₅₇	CH ₃ antisymmetric stretch	3152
V ₅₈	C(3) + NH ₂ antisymmetric stretch	3254
V ₅₉	C(3) + NH ₂ symmetric stretch	3372
V ₆₀	NH ₂ antisymmetric stretch	3532

Table S4. Cartesian coordinates of the atoms in [MAT][DCA] and [MAT][DCA] reacted with nitrogen dioxide calculated at the M06/6-31+G(d,p) level of theory.

structure	coordinates (Å)			
[MAT][DCA] 	C	0.944	0.020	-0.821
	C	2.064	-0.583	0.940
	N	1.357	1.077	-0.152
	H	0.364	0.012	-1.736
	H	2.530	-1.234	1.667
	C	1.162	2.484	-0.467
	H	0.605	2.555	-1.402
	H	0.593	2.950	0.340
	H	2.142	2.955	-0.570
	N	2.058	0.721	0.958
	N	1.388	-1.056	-0.146
	N	1.215	-2.407	-0.436
	H	0.209	-2.611	-0.360
	H	1.537	-2.574	-1.388
	N	-2.959	-0.119	0.288
	C	-2.254	-1.196	0.134
C	-2.444	1.069	0.144	
N	-1.688	-2.222	0.009	
N	-2.057	2.175	0.031	
[MAT][DCA] + NO ₂ 	C	1.277	-0.354	0.345
	C	3.410	-0.356	-0.085
	N	1.605	-1.404	-0.380
	H	0.273	-0.090	0.671
	H	4.440	-0.028	-0.087
	C	0.731	-2.448	-0.893
	H	-0.268	-2.298	-0.476
	H	0.706	-2.384	-1.983
	H	1.127	-3.417	-0.581
	N	2.937	-1.427	-0.660
	N	2.416	0.333	0.547
	N	2.606	1.535	1.224
	H	1.982	2.230	0.798
	H	2.356	1.399	2.202
	N	-1.914	2.295	-0.741
	C	-0.636	2.554	-0.576
	C	-2.430	1.172	-0.397
	N	0.495	2.840	-0.454
	N	-1.883	0.098	0.116
	N	-2.749	-0.932	0.407
O	-3.958	-0.833	0.228	
O	-2.208	-1.941	0.857	

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